

Homework 4

1) Activity = λN
 decay constant λ \uparrow N \leftarrow # of atoms present

Since the Ruthenium isotope is not expected to occur in pure form in a random sample, we'll assume it's all created by the Technetium

So, the number of $^{99}_{43}\text{Tc}$ we started with will be

$$\underbrace{\#^{99}_{43}\text{Tc}_{\text{now}}}_{N(t)} + \#^{99}_{44}\text{Ru}_{\text{now}} = N(0)$$

$$^{99}_{43}\text{Tc} \text{ decays like } N(t) = N(0)e^{-\lambda t}$$

$$\text{so } \ln\left(\frac{N(t)}{N(0)}\right) = -\lambda t$$

$$t = \ln\left(\frac{N(t)}{N(0)}\right) / -\lambda$$

$$\frac{N(t)}{N(0)} = \frac{\frac{\#^{99}_{43}\text{Tc}_{\text{now}}}{t_{0t}}}{\frac{\#^{99}_{43}\text{Tc}_{\text{now}}}{t_{0t}} + \frac{\#^{99}_{44}\text{Ru}_{\text{now}}}{t_{0t}}} = \frac{0.012}{0.012 + 2.2}$$

$$\lambda = \frac{\text{Activity}}{N} \approx \frac{1200/\text{s}}{(6.02 \times 10^{23}/^{99}\text{g}) \cdot 1 \times 10^{-6}\text{g}} = 1.973 \times 10^{-13}/\text{s}$$

$$t = \ln\left(\frac{0.012}{0.012 + 2.2}\right) / (-1.93 \times 10^{-13}/\text{s}) = 2.7 \times 10^{13}\text{s}$$

$$= 8.57 \times 10^5 \text{ years (this is a slightly used model!)}$$

note: we can't make the other components from Tc

- i) The difference in number of nucleons must be divisible by 4 for α decay
- ii) can't make higher A unless there's a beam or source of big nuclei
- iii) Can make the smaller A by fission I guess, but we should see the other products!

2) look at BE. for $A \neq Z$ # of nucleons

$$BE \text{ for } {}_Z^{A-1}X = (A-1)m(n)c^2 + Zm(H)c^2 - m({}_Z^{A-1}X)c^2$$

$$\text{for } {}_Z^AX = Am(n)c^2 + Zm(H)c^2 - m({}_Z^AX)c^2$$

Binding energy for one neutron and ${}_Z^{A-1}X$

$$BE = m(n)c^2 + m({}_Z^{A-1}X)c^2 - m({}_Z^AX)c^2 = S_n$$

this is the energy we need to supply to knock out that neutron

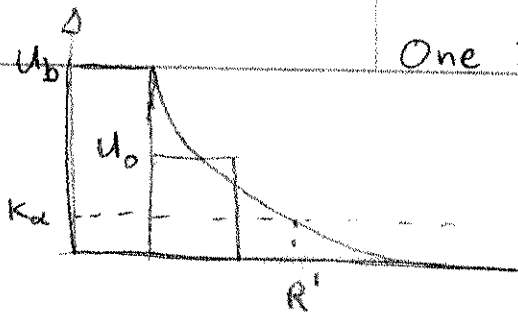
$$\begin{aligned} {}^{17}_8\text{O} \quad S_n &= m(n)c^2 + m({}^{16}_8\text{O})c^2 - m({}^{17}_8\text{O})c^2 \\ &= (1.008664924u + 15.994915u - 16.999132u) \frac{931.5 \text{ MeV}}{u} \\ &= 4.14 \text{ MeV} \end{aligned}$$

$$\begin{aligned} {}^7_3\text{Li} \quad S_n &= (1.008664924u + 6.015122u - 7.016004u) \frac{931.5}{u} \\ &= 7.24 \text{ MeV} \end{aligned}$$

$$\begin{aligned} {}^{56}_{26}\text{Fe} \quad S_n &= (1.008664924u + 55.934924u - 56.935398u) \frac{931.5 \text{ MeV}}{u} \\ &= 7.63 \text{ MeV} \end{aligned}$$

One Interpretation

3)



$$U_0 = \frac{1}{2}(U_b + K_\alpha), U_0 - E = \frac{1}{2}(U_b - K_\alpha)$$

$$L = \frac{1}{2}(R' - R)$$

want $t_{1/2}$ for ^{232}Th & ^{218}Th

$$\left\{ U_b = 30 \text{ MeV} \right\} \left\{ K_\alpha = 4.01 \text{ MeV}, 9.85 \text{ MeV} \right.$$

$$\left. \left\{ R = (1.2 \times 10^{-15} \text{ m}) A^{1/3} \right. \right.$$

$$\left. \left. = 7.37 \text{ fm}, 7.22 \text{ fm} \right. \right.$$

measured

$$^{232}\text{Th } t_{1/2} = 1.4 \times 10^{10} \text{ y}$$

$$^{218}\text{Th } t_{1/2} = 1.1 \times 10^{-7} \text{ s}$$

$$\lambda = \frac{v}{2R} e^{-2KL}$$

$$t_{1/2} = \frac{\ln(2)}{\lambda}$$

$$= \frac{\ln(2) 2R}{v} e^{2KL}$$

$$R' = \left(\frac{K_\alpha^2 (2)(Z-2)}{K_\alpha} \right)$$

$$= \frac{(9 \times 10^9 \text{ Nm}^2/\text{C}^2) (1.6 \times 10^{-19} \text{ C})^2 (2)(90-2)}{4.01 \times 10^6 (1.6 \times 10^{-19} \text{ C}) \text{ V}}$$

$$= 6.32 \times 10^{-14} \text{ m } ^{232}\text{Th}$$

$$= 6.32 \times 10^{-14} \text{ m} \cdot \frac{4.01}{9.85} = 2.57 \times 10^{-14} \text{ m}$$

for ^{218}Th

$$^{232}\text{Th } L = \frac{1}{2}(63.2 - 7.37) \text{ fm} = \frac{55.83 \text{ fm}}{2}$$

$$^{218}\text{Th } L = \frac{1}{2}(25.7 - 7.22) \text{ fm} = \frac{18.48 \text{ fm}}{2}$$

$$v = \left(\sqrt{2K_\alpha / mc^2} \right) c$$

$$^{232}\text{Th } = \left(\sqrt{\frac{2(4.01 \text{ MeV})}{3727.3803 \text{ MeV}}} \right) c = 1.39 \times 10^7 \text{ m/s}$$

$$^{218}\text{Th } = 2.18 \times 10^7 \text{ m/s}$$

$$K = \sqrt{\left(\frac{2m}{\hbar^2} \right) (U_0 - E)} = \sqrt{\frac{8\pi^2 mc^2}{(hc)^2} \left(\frac{1}{2}(U_b - K_\alpha) \right)}$$

$$^{232}\text{Th } K = \sqrt{\frac{8\pi^2 (3727.3803 \text{ MeV})}{(1240 \times 10^{-6} \text{ MeV nm})^2} \cdot \frac{1}{2} (30 \text{ MeV} - 4.01 \text{ MeV})}$$

$$= 1.58 \times 10^6 / \text{nm} = 1.58 / \text{fm}$$

$$^{218}\text{Th } K = 1.39 \times 10^6 / \text{nm} = 1.39 / \text{fm}$$

$$^{232}\text{Th } t_{1/2} = \frac{\ln(2) (2) (7.37 \times 10^{-15} \text{ m})}{1.39 \times 10^7 \text{ m/s}} e^{2(1.58 / \text{fm})(27.92 \text{ fm})} = 4.83 \times 10^9 \text{ y}$$

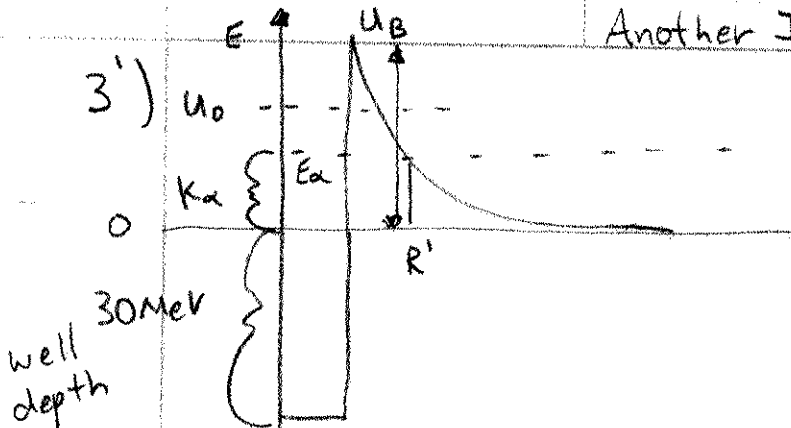
not too bad

$$^{218}\text{Th } t_{1/2} = \frac{\ln(2) (2) (7.22 \times 10^{-15} \text{ m})}{2.18 \times 10^7 \text{ m/s}} e^{2(1.39)(9.24)} = 6.57 \times 10^{-11} \text{ s}$$

bit off!

see class notes for "better" job
but $R/R' \sim 1$ for ^{218}Th !

Another Interpretation



$$U_b = \frac{1}{4\pi\epsilon_0} \frac{q^2 (Z)(Z-2)}{R_{nuc}}$$

$$v = \sqrt{(30 \text{ MeV} + K_\alpha)^2 / m}$$

$$232 \quad U_b = K_\alpha \frac{R'}{R_{nuc}} = 4.01 \text{ MeV} \left(\frac{63.2}{7.37} \right) = 34.39 \text{ MeV}$$

$$218 \quad U_b = 9.85 \text{ MeV} \left(\frac{25.7}{7.22} \right) = 35.06 \text{ MeV}$$

He nucleus will be moving faster inside here (just like a rocket having v_{esc} at earth's surface has $v=0$ just outside the earth's potential well

if U_0 is midway between U_b & E_α

very deceptive part of problem!

$$U_0 = \frac{(30 \text{ MeV} + U_b) + (30 \text{ MeV} + K_\alpha)}{2}$$

$$= 30 \text{ MeV} + \frac{1}{2}(U_b + K_\alpha)$$

$$E = 30 \text{ MeV} + K_\alpha$$

$$U_0 - E = \frac{1}{2}(U_b - K_\alpha)$$

$$232 \quad v = c \sqrt{(34 \text{ MeV})^2 / 3727.3803 \text{ MeV}} = 4.05 \times 10^7 \text{ m/s}$$

$$218 \quad v = c \sqrt{39.85 \text{ MeV} (2) / 3727.38 \text{ MeV}} = 4.39 \times 10^7 \text{ m/s}$$

$$232 \quad k = \frac{1.58}{\text{Fm}} \cdot \sqrt{\frac{34.39 - 4.01}{30 - 4.01}} = \frac{1.70}{\text{Fm}}$$

$$218 \quad k = \frac{1.39}{\text{Fm}} \sqrt{\frac{35.06 - 9.85}{30 - 9.85}} = \frac{1.55}{\text{Fm}}$$

$$232 \quad t_{1/2} = \frac{\ln(2) (2) (7.37 \times 10^{-15} \text{ m})}{4.05 \times 10^7 \text{ m/s}} e^{\frac{2(1.7 \text{ Fm})(27.92 \text{ Fm})}{3.15 \times 10^8 \text{ s/y}}} = 1.35 \times 10^{12} \text{ y}$$

$$218 \quad t_{1/2} = \frac{\ln(2) (2) (7.22 \times 10^{-15} \text{ m})}{4.39 \times 10^7 \text{ m/s}} e^{2(1.55)(9.24)} = 6.27 \times 10^{-10} \text{ s}$$

to do a better job, we really need to know the shape of the potential. (then we could do tricks like in class)

232 $t_{1/2} = 1.2 \times 10^{12} \text{ y}$
 class 218 $t_{1/2} = 2.2 \times 10^{-6} \text{ s}$

4) Rate of decay goes like # of Atoms that can decay

$$R(t) = R_0 e^{-\lambda t} = R_0 e^{-\ln(2) t / t_{1/2}}$$

$$R_0 = \lambda N_0$$

$$\frac{R(0)}{R(t)} = \frac{R_0}{R(t)} = e^{\ln(2) t / t_{1/2}}$$

$$\ln\left(\frac{R_0}{R}\right) = \ln(2) t / t_{1/2}$$

$$t = t_{1/2} \frac{\ln\left(\frac{R_0}{R}\right)}{\ln(2)} = 5730 \text{ y} \frac{\ln\left(\frac{12.4}{3.5}\right)}{\ln(2)}$$

$$= 1.05 \times 10^4 \text{ y} \quad (3.3 \times 10^3 \text{ y})$$

(can also ask how many half lives is

this $\left(\frac{1}{2}\right)^N = \frac{3.5}{12.4} \quad N = 1.825$

$$t = N(t_{1/2}) = \text{same}$$