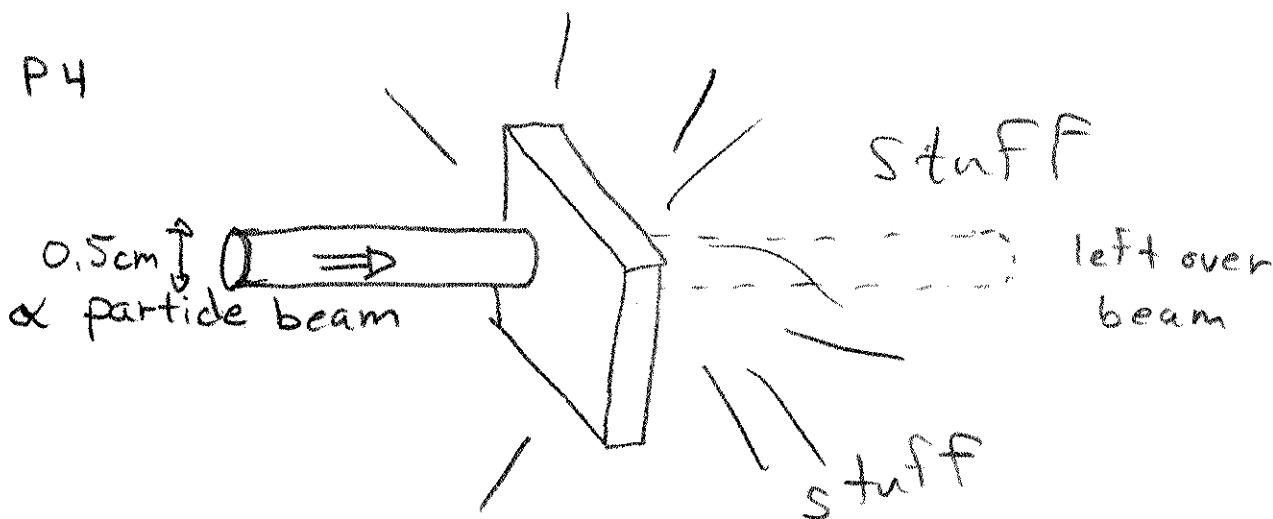


Homework 5

Krane

C.13 P4, 8, 15, 21, 31

P4



recall our scattering expression from class

$$\frac{\# \text{ scattered particles}}{s} = \left(\text{Probability of scattering} \right) \frac{\# \text{ incident particles}}{s}$$

$$\& \text{ the Probability} = \frac{\text{effective area}}{\text{beam area}}$$

$$\text{effective area} = \underbrace{\sigma}_{\text{cross section}} \underbrace{\frac{N}{V}}_{\# \text{ of nuclei/Volume}} \underbrace{t}_{\text{target thickness}} \underbrace{A}_{\text{beam area}}$$

For ^{63}Cu density is $8.92 \times 10^3 \frac{\text{kg}}{\text{m}^3} \cdot \left(\frac{63}{(63 \cdot 692 + 65 \cdot 308)} \right)$
 $= 8.83 \times 10^3 \frac{\text{kg}}{\text{m}^3}$ my estimate based on abundance

$$\frac{N}{V} = \left(\frac{6.02 \times 10^{23}}{\text{mole}} \right) \cdot \frac{1 \text{ mole}}{0.063 \text{ kg}} \cdot 8.83 \times 10^3 \frac{\text{kg}}{\text{m}^3} = 8.44 \times 10^{28} / \text{m}^3$$

$$\frac{\# \text{ incident particles}}{s} = I_\alpha / \text{charge of } \alpha = \left(\frac{7.5 \times 10^{-6} \text{ C}}{s} \right) / (2 \cdot 1.6 \times 10^{-19} \text{ C}) = 2.34 \times 10^{13} / s$$

$$\text{so, } \frac{\# \text{ neutrons emitted}}{s} = \left(\underbrace{1.25 \times 10^{-28}}_{\sigma} \right) \left(\underbrace{8.44 \times 10^{28}}_{N/V} / \text{m}^3 \right) \left(\underbrace{2.5 \times 10^{-6}}_t \text{ m} \right) \left(\underbrace{2.34 \times 10^{13}}_s / s \right) = 6.17 \times 10^8 / s$$

assumes no absorption in target & no attenuation of the beam (good assumption, not much stuff comes out relatively)

The rate at which an isotope is produced

$$R = \left(\text{Probability of interaction} \right) \times \text{Rate of incident beam}$$

assuming the beam covers the whole sample \Rightarrow

$$= \left(\frac{\text{\# of nuclei present in beam}}{\text{mass of sample}} \cdot \frac{\text{molar mass of sample}}{\text{molar mass of sample}} \right) \sigma_I \cdot \frac{\text{\# incident beam particles}}{s \text{ (Area of beam)}} = \text{beam Flux, } \phi \cdot \text{interaction cross section}$$

The number of isotopes present after an irradiation time t_0 is

$$N(t_0) = \frac{R}{\lambda} (1 - e^{-\lambda t_0}) \quad \text{like a charging up capacitor}$$

if the activity is $\lambda N(t_0) = R (1 - e^{-\lambda t_0})$

$$\lambda N(t_0) = \left(N_A \frac{m}{M} \right) \sigma_I \phi (1 - e^{-\lambda t_0})$$

$$m = \frac{M \lambda N(t_0)}{N_A \sigma_I \phi} \cdot \frac{1}{(1 - e^{-\lambda t_0})}$$

${}^{60}_{Co} \lambda = \frac{\ln(2)}{(5.27y) \left(\frac{31.5Ms}{y} \right)}$
 $= 4.175 \times 10^{-9} / s$
 ${}^{51}_{Ti} \lambda = \frac{\ln(2)}{5.8m \cdot \frac{60s}{m}}$
 $= 1.99 \times 10^{-3} / s$

For ${}^{60}_{Co}$ λt_0 is tiny, so we can Taylor expand $= 1.99 \times 10^{-3} / s$

$$m = \frac{M(\lambda N(t_0))}{N_A \sigma_I \phi} \cdot \frac{1}{\lambda t_0} \left\{ \lambda t_0 = (4.175 \times 10^{-9} / s) (150s) \right\}$$

$$= \frac{(59g)}{\text{mole}} \left(\frac{12}{s} \right) / \frac{6.02 \times 10^{23}}{\text{mole}} \cdot 1.9 \times 10^{-28} m^2 \cdot \frac{3.0 \times 10^{16}}{s \cdot m^2} \cdot 6.26 \times 10^{-7}$$

$$= 3.3 \mu g$$

For ${}^{50}_{Ti}$ (5.25% of Ti present)

$$m = \frac{M(\lambda N(t_0))}{N_A \sigma_I \phi} \cdot \frac{1}{(1 - e^{-\lambda t_0})} = \frac{\left(\frac{50g}{\text{mole}} \cdot \frac{105}{s} \right)}{\frac{6.02 \times 10^{23}}{\text{mole}} \cdot 1.4 \times 10^{-29} m^2 \cdot \frac{3.0 \times 10^{16}}{s \cdot m^2} (1 - e^{-0.2985})}$$

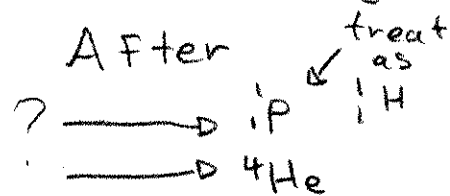
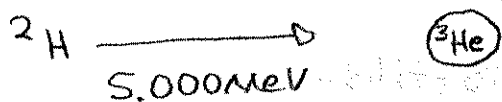
$$= 8.05 \times 10^{-8} g$$

mass of Ti present = $\frac{8.05 \times 10^{-8} g}{.0525} = 1.53 \mu g$

P 15

Energy is conserved
Momentum is conserved
before

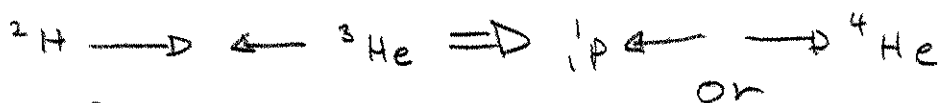
^2H is moving non-relativistically



Energy $m(^2\text{H})c^2 + m(^3\text{He})c^2 + \underbrace{5.000\text{MeV}}_{KE_0} = m(^1\text{H})c^2 + KE_p + m(^4\text{He})c^2 + KE_{He}$

Momentum $m(^2\text{H})v_0 = p_0 = \underbrace{m(^1\text{H})v_p}_{p_p} + \underbrace{m(^4\text{He})v_{He}}_{p_{He}}$

Think about this a second. If this collision takes place in the center of mass, we can have one of 2 situations



lets get some stuff out of the way: $^4\text{He} \leftarrow \rightarrow \text{ } ^1\text{H}$ or

$Q = (2.014102u + 3.016029u - 1.007825u - 4.002603u) \frac{931.5\text{MeV}}{u}$
 $= 18.35\text{ MeV}$ (lots of extra energy)

so $23.35\text{ MeV} = KE_p + KE_{He} = \frac{p_p^2}{2m_p} + \frac{p_{He}^2}{2m_{He}}$

$\sqrt{2m_{He} KE_0} = p_0 = 136.97 \frac{\text{MeV}}{c} = p_p + p_{He}$

$p_{He}c = 136.97\text{MeV} - p_p c$

$23.35 = \frac{p_p^2 c^2}{2m_p c^2} + \frac{(136.97 - p_p c)^2}{2m_{He} c^2}$ all in MeV

mult both sides by $2m_p c^2$

or $0 = 1.25179 p_p^2 c^2 - 68.975 p_p c + 39,150.8$

$p_p c = \frac{68.975 \pm \sqrt{68.975^2 + 4(1.25179)(39,150.8)}}{2(1.25179)}$

$= 206.53\text{ MeV}, 151.43\text{ MeV} \Rightarrow KE_p = 22.72\text{ MeV}, 12.2\text{ MeV}$

so $KE_{\alpha} = 23.35\text{ MeV} - KE_p = 0.63\text{ MeV}, 11.1\text{ MeV}$

P21

calculate neutron separation energies \leftrightarrow excitation energies

$${}^{236}_{92}\text{U} \quad S_n = [1.008665u + 235.043922u - 236.045561u] \frac{931.5 \text{ MeV}}{u}$$

$$= 6.54 \text{ MeV} = \text{energy available to excite } {}^{236}_{92}\text{U}$$

$${}^{238}_{92}\text{U} \quad S_n = [1.008665u + 238.050784u - 239.054289u] \frac{931.5 \text{ MeV}}{u}$$

$$= 4.81 \text{ MeV} = \text{energy available to excite } {}^{238}_{92}\text{U}$$

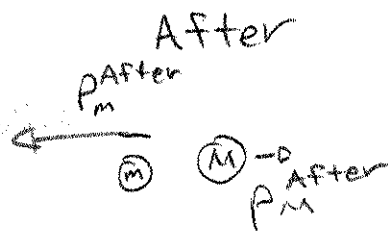
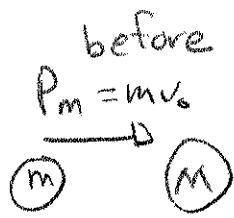
If we need about 6.54 MeV to initiate fission in a large nucleus, ${}^{238}_{92}\text{U}$ will need an additional $\sim 1.7 \text{ MeV}$ of energy to get there

$${}^{239}_{94}\text{Pu} \quad S_n = [1.008665u + 239.052156u - 240.053807u] \frac{931.5 \text{ MeV}}{u}$$

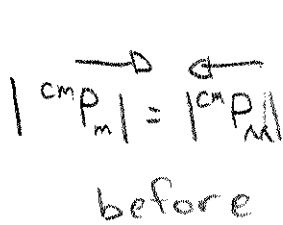
$$= 6.53 \text{ MeV}$$

has a high excitation energy too, probably has fission at low neutron energies.

P31



in center of momentum frame, particles



just reverse direction
but keep same $|P|$

$$mv_0 = (m+M)v_{\text{cm}}$$

before $p_m^{\text{cm}} = m(v_0 - v_{\text{cm}}^{\text{before}})$, $p_M^{\text{cm}} = M(-v_{\text{cm}})$

after $p_m^{\text{cm}} = m(v_{\text{cm}}^{\text{after}} - v_0)$, $p_M^{\text{cm}} = M(v_{\text{cm}})$

translate to lab frame for

$$v^{\text{After}} = v_{\text{lab}} = v_{\text{after}}^{\text{cm}} + v_{\text{cm}} = 2v_{\text{cm}} - v_0$$

$$= \frac{2m(v_0)}{m+M} - v_0 = \frac{2m(v_0) - (m+M)v_0}{m+M}$$

$$= \frac{m-M}{m+M} v_0$$

$$\Delta KE = \frac{1}{2} m v_0^2 - \frac{1}{2} m \left(\frac{m-M}{m+M} \right)^2 v_0^2$$

$$= \frac{1}{2} m v_0^2 \left(\frac{(m+M)^2 - (m-M)^2}{(m+M)^2} \right)$$

$$= KE_0 \frac{4mM}{(m+M)^2} = KE_0 \frac{4m/M}{(1+m/M)^2}$$