

Suppose that you have a decay chain.

Particle₁ → Particle₂

Particle₂ → X
↳ detect this

$$\frac{dN_2}{dt} = \lambda_1 N_1 - \lambda_2 N_2$$

↑
rate at which
N₂ appears

↑
rate at which
N₂ disappears

guess a solution $N_2(t) = A e^{-\lambda_1 t} + B e^{-\lambda_2 t}$

$$\text{@ } t=0 \quad N_2(0) = 0 \quad A = -B$$

$$N_2(t) = A(e^{-\lambda_1 t} - e^{-\lambda_2 t})$$

$$\frac{dN_2}{dt} = A(-\lambda_1 e^{-\lambda_1 t} + \lambda_2 e^{-\lambda_2 t})$$

$$= \lambda_1 N_{10} e^{-\lambda_1 t} - \lambda_2 (A)(e^{-\lambda_1 t} - e^{-\lambda_2 t})$$

group like terms

simplify

$$A(-\lambda_1) e^{-\lambda_1 t} = \lambda_1 N_{10} e^{-\lambda_1 t} - \lambda_2 A e^{-\lambda_1 t}$$
$$A \lambda_2 e^{-\lambda_2 t} = A \lambda_2 e^{-\lambda_2 t}$$

$$-A \lambda_1 = \lambda_1 N_{10} - \lambda_2 A$$

$$A(\lambda_2 - \lambda_1) = \lambda_1 N_{10}$$

$$A = \frac{\lambda_1 N_{10}}{\lambda_2 - \lambda_1}$$

$$N_2(t) = \left(\frac{\lambda_1}{\lambda_2 - \lambda_1} \right) N_{10} (e^{-\lambda_1 t} - e^{-\lambda_2 t})$$

if you have some N₂ to start with

$$N_2(t) = \left(\frac{\lambda_1}{\lambda_2 - \lambda_1} \right) N_{10} (e^{-\lambda_1 t} - e^{-\lambda_2 t}) + N_{02} e^{-\lambda_2 t}$$

Last time we talked about binding energy in a nucleus and that it has some general properties due to the nature of the strong force, coulomb repulsion of the protons, and the fact that the neutrons and protons are fermions.

1st, lets consider larger nuclei where things are likely to work better on average. Big is about $A=20$ for our purpose.

Next, consider what it must look like in a big nucleus

The nucleons on the outside have less bonds



The nucleons on the inside has lots of "bonds"

assume # bonds \propto binding energy

$$\text{Binding energy} \propto \# \text{ nucleons on the inside} \times \# \text{ inside bonds} \\ + \# \text{ nucleons on the outside} \times \# \text{ outside bonds}$$

or $\# \text{ nucleons} (\# \text{ bonds on the inside}) - \text{correction for nucleons on the surface}$

$$\text{Binding Energy} \propto a_1 A - a_2 A^{2/3}$$




\nearrow interior binding coefficient \nearrow surface correction term

recall that nucleons are closely packed & the formula $r = r_0 A^{1/3}$ works pretty well for $r_0 = 1.25 \text{ fm}$.

Now we need a way to deal with the coulomb repulsion. This we can add in by considering the potential energy of a collection of charged particles in a pairwise fashion (neglecting different r 's, well combine everything in $\frac{1}{r_{avg}} \propto \frac{1}{A^{1/3}}$)

how many ways can you choose 2 distinct pairs of items from a collection of Z distinct objects \rightarrow $\frac{Z!}{2!(Z-2)!}$ $\leftarrow N!$ ways to choose N items
(need to be distinct ordered pair)

Or you can just work it out

	# Pairwise	# protons	
	1	2	
	3	3	get 2 more
	6	4	get 3 more
	10	5	get 4 more
	⋮	⋮	

or $1 + 3 + 6 + 10 + \dots$

$$\frac{1}{2} Z(Z-1) \quad Z$$

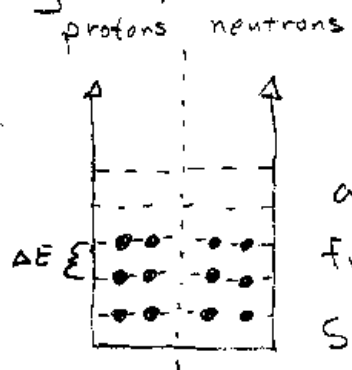
$$\sim \frac{Z^2}{2}$$

so our binding energy becomes

$$E_{bind} \propto a_1 A - a_2 A^{2/3} - a_c \frac{Z^2}{A^{1/3}}$$

The next term we have to worry about can be thought of 2 ways. One is a simple energy argument, one is using our Fermi gas model.

Consider a very simple model of energy levels in a nucleus

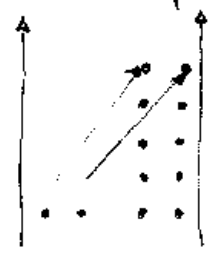


recall that we have a mechanism to change from protons \leftrightarrow neutrons. So, if we pile on lots of neutrons, for instance, it is likely some of them will convert into protons to get a higher binding energy. (deeper in the well)

Consider the energy cost to promote 2 protons to become neutrons.



$2\Delta E$ ξ to promote 2 more



$2(3\Delta E)$

ξ to promote 2 more $2(5\Delta E)$

$(\# \text{ proton pairs})$	$(\# \text{ protons}) \text{ promoted}$	Energy cost
1	2	$2\Delta E$
2	4	$2\Delta E + 6\Delta E = 8\Delta E$
3	6	$18\Delta E$
4	8	$32\Delta E$
\vdots	\vdots	
X	X	$2(\# \text{ proton pairs})^2$
$\frac{1}{4}$		$\# \text{ proton pairs} = (\# \text{ neutrons} - \# \text{ protons}) \frac{1}{4}$